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# 3-Amino-1-methyl-9,10-dihydro-phenanthrene-2,4-dicarbonitrile 

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Key indicators: single-crystal X-ray study; $T=100 \mathrm{~K}$; mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$; $R$ factor $=0.034 ; w R$ factor $=0.091$; data-to-parameter ratio $=7.4$.

The asymmetric unit of the title compound, $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{~N}_{3}$, contains two independent molecules, which are non-planar as they are buckled owing to the ethylene portion. The dihedral angle between the benzene rings is $26.4(1)^{\circ}$ in one molecule and $32.9(1)^{\circ}$ in the other. In the crystal, the molecules are disposed about a false inversion center, and are linked by two $\mathrm{N}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds, generating a dimer. The dimers are linked by further $\mathrm{N}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds, resulting in a chain that runs along the longest axis of the orthorhombic unit cell.

## Related literature

For the synthesis of dihydrophenanthrenes, see: Dellagreca et al. (2000); Ram \& Goel (1997).


## Experimental

Crystal data
$\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{~N}_{3}$
$M_{r}=259.30$
Orthorhombic, $\mathrm{Pna}_{1}$
$a=26.8587$ (7) $\AA$
$b=8.8158$ (2) $\AA$
$c=11.2035$ (3) A
$V=2652.78(12) \AA^{3}$
$Z=8$
$C u K \alpha$ radiation
$\mu=0.62 \mathrm{~mm}^{-1}$
$T=100 \mathrm{~K}$
$0.30 \times 0.20 \times 0.02 \mathrm{~mm}$

Data collection
Agilent SuperNova Dual diffractometer with an Atlas detector
Absorption correction: multi-scan (CrysAlis PRO; Agilent, 2010) $T_{\text {min }}=0.836, T_{\text {max }}=0.988$

## Refinement

$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.034$
$w R\left(F^{2}\right)=0.091$
$S=1.09$
2800 reflections
379 parameters
1 restraint

H atoms treated by a mixture of independent and constrained refinement
$\Delta \rho_{\max }=0.16 \mathrm{e} \AA^{-3}$
$\Delta \rho_{\min }=-0.20 \mathrm{e}^{-3}$

Table 1
Hydrogen-bond geometry ( $\AA,{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{~N} 2-\mathrm{H} 21 \cdots \mathrm{~N} 4$ | $0.91(4)$ | $2.15(4)$ | $3.007(3)$ | $156(3)$ |
| $\mathrm{N} 2-\mathrm{H} 22 \cdots \mathrm{~N} 6^{\mathrm{i}}$ | $0.91(3)$ | $2.38(3)$ | $3.265(3)$ | $164(2)$ |
| $\mathrm{N} 5-\mathrm{H} 51 \cdots \mathrm{~N} 1^{i}$ | $0.91(4)$ | $2.12(4)$ | $3.012(3)$ | $168(3)$ |
| $\mathrm{N} 5-\mathrm{H} 52 \cdots \mathrm{~N} 3$ | $0.91(3)$ | $2.41(3)$ | $3.283(3)$ | $161(3)$ |

Symmetry codes: (i) $x, y, z+1$; (ii) $x, y, z-1$.

Data collection: CrysAlis PRO (Agilent, 2010); cell refinement: CrysAlis PRO; data reduction: CrysAlis PRO; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: $X-S E E D$ (Barbour, 2001); software used to prepare material for publication: publCIF (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5310).

## References

Agilent (2010). CrysAlis PRO. Agilent Technologies, Yarnton, England.
Barbour, L. J. (2001). J. Supramol. Chem. 1, 189-191.
Dellagreca, M., Fiorentino, A., Monaco, P., Previtera, L. \& Zarrelli, A. (2000). J. Chem. Ecol. 26, 587-600.

Ram, V. J. \& Goel, A. (1997). J. Chem. Res. pp. 460-461.
Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
Westrip, S. P. (2010). J. Appl. Cryst. 43, 920-925.

